



dig
Dolan Integration Group

Geochemistry for Energy

11025 Dover Street Unit 800
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p: 303.531.2030

**Quantitative High Resolution Whole Oil Gas Chromatography
Analytical Results**

Job #: 260315541

Lab #: DIG-043376

Client: Energy and Carbon Management Commission

Project/ Well Name: FAC 5 GAS LINE PRODUCT

API:

Facility ID:

Sample Types: Oil

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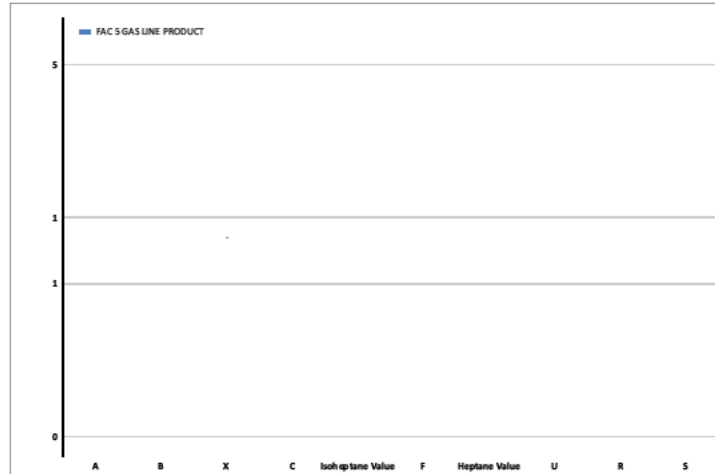
Definitions & Interpretations

Thompson ratios describe processes affecting light hydrocarbons (C₆-C₇)

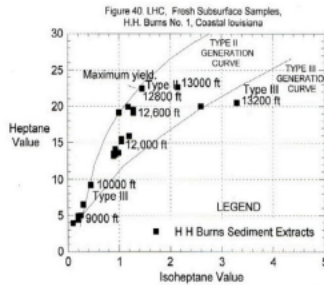
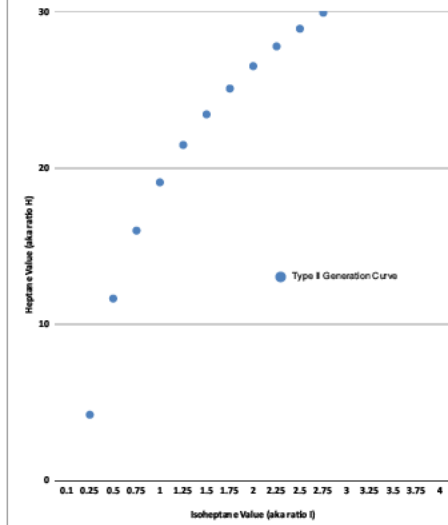
Name	Ratio	Property	Process
A	benzene / n -hexane	Aromaticity	Fractionation, water washing, TSR
B	toluene / n -heptane	Aromaticity	Fractionation, water washing, TSR
X	(m-xylene + p-xylene) / n-octane	Aromaticity	Fractionation, water washing, TSR
C	(n -hexane + n -heptane) / (cyclohexane + methylcyclohexane)	Paraffinicity	Maturity, biodegradation
I	(2- + 3-methylhexane) / (1c3- + 1t3- + 1t2-DMCP)	Paraffinicity	Maturity, source, biodegradation
F	n -heptane / methylcyclohexane	Paraffinicity	Maturity, biodegradation
H	(100*n -heptane) / (Σcyclohexane + ΣC ₇ HCs)	Paraffinicity	Maturity, source, biodegradation
S	n -hexane / 2,2-dimethylbutane	Paraffin branching	Maturity, source, biodegradation
R	n -heptane / 2-methylhexane	Paraffin branching	Maturity, source, biodegradation
U	cyclohexane / methylcyclohexane	Naphthene branching	Maturity, source

DMCP, dimethylcyclopentane; H, heptane value; I, isoheptane value; Σcyclohexane, [cyclohexane + 1,1-DMCP + 1c3-DMCP + 1t3-DMCP + 1t2-DMCP + methylcyclohexane]; ΣC₇ HCs, [Σcyclohexane + 2-methylhexane + 3-methylhexane + n -heptane + 2,3-dimethylpentane + 3-ethylpentane]; 1c3, 1-cis-3; 1t3, 1-trans-3.

Thompson Ratios



Heptane Plot - Light Hydrocarbon Maturity



Thompson K F M. Classification and thermal history of petroleum based on light hydrocarbons. *Geochimica et Cosmochimica Acta*, 1983, 47(2):303-316.

Oil Alteration Vectors

Modified from Thompson, KFM 1987 (Fig. 12)

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Evaporative Fractionation

- Physical processes can also fractionate petroleum during migration.
- As petroleum migrates upward, temperature and pressure decrease. If the bubble point is reached, a single phase fractionates into liquid and gas phases that can migrate separately.
- Partitioning of petroleum between these phases depends on the vapor equilibrium constants for each component (Thompson, 1987). This can result in reservoirs filled with petroleum from a common source, but with different gross compositions.
- The effects of evaporative or phase fractionation were described based mostly on laboratory experiments (e.g. Thompson, 1987; Larter and Mills, 1991; van Graas et al., 2000).
- The process leads to oils deficient in light ends and enriched in aromatics, as exemplified by many U.S. Gulf Coast oils (Thompson, 1987, 1988).

Peters & Fowler, *Organic Geochemistry* 33 (2002) pp 5-36

CONNAN PLOT - SOURCE ROCK DEPOSITION/MATURATION CHARACTERIZATION

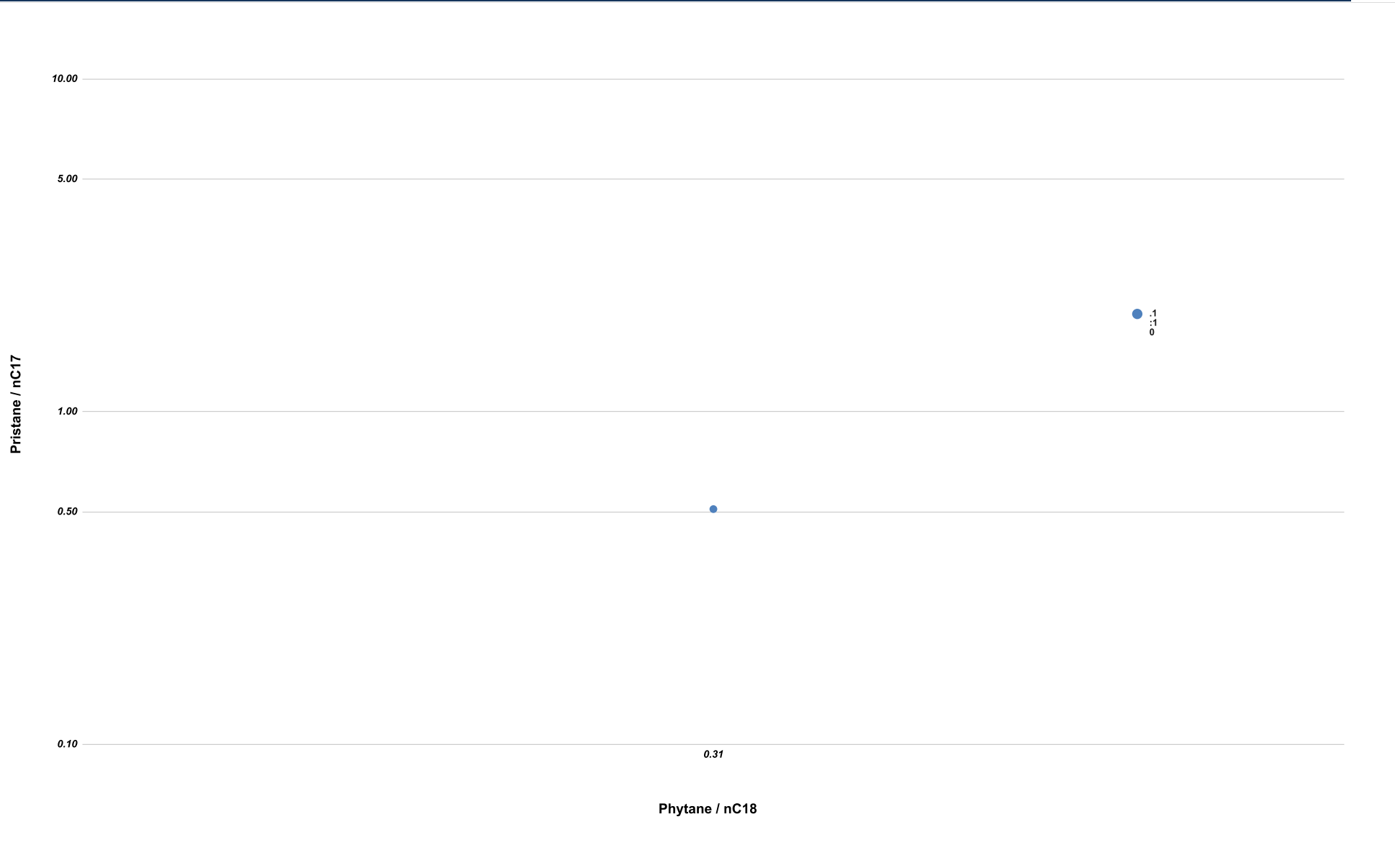


Table 7.3. Halpern (1995) C₇ ratios for use in star diagrams to differentiate oils

Name	Ratio	ΔBP (°C)	ΔSolubility (ppm)	Process
TR1	Toluene/X	22.8	496	Water washing ↑ Biodegradation Evaporation
TR2	nC ₇ /X	10.6	-21.8	
TR3	3-Methylhexane/X	4.0	-21.4	
TR4	2-Methylhexane/X	2.2	-21.5	
TR5	P2/X	(3.2)	(-21.4)	
TR6	1-cis-2-Dimethylcyclopentane/X	11.7	-11.0	
TR7	1-trans-3-Dimethylcyclopentane/X	3.0	-4.0	
TR8	P2/P3	(6)	(-2.4)	
C1	2,2-Dimethylpentane/P3	(-5.8)	(-0.6)	↑ Correlation ↓
C2	2,3-Dimethylpentane/P3	(4.8)	(0.3)	
C3	2,4-Dimethylpentane/P3	(-4.5)	(-0.6)	
C4	3,3-Dimethylpentane/P3	(1.1)	(0.9)	
C5	3-Ethylpentane/P3	(8.5)	(-2.0)	

X = 1,1-dimethylcyclopentane, boiling point 87.8°C, solubility 24 ppm. P2 = 2-methylhexane + 3-methylhexane, boiling point 91°C, solubility 2.6 ppm. P3 = 2,2-dimethylpentane + 2,3-dimethylpentane + 2,4-dimethylpentane + 3,3-dimethylpentane + 3-ethylpentane, boiling point 85°C, solubility 5 ppm.

ΔBP = boiling point numerator – boiling point denominator (°C).

ΔSolubility = solubility of numerator – solubility of denominator (ppm in distilled water).

Parentheses indicate average values for mixtures.

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Turnaround Time**: Standard (≤ 10 Business days) Rush (≤ 5 Business days) Expedited Rush (≤ 3 Business days)

Container Number	Sample Identification	Date Sampled	Time	Sample Type*	Gas Composition	d13C of Methane (C1)	d13C of Ethane (C2)	d13C of Propane+ (C3+)	d13C of Carbon Dioxide (CO2)	dD of Methane (C1)	Whole Oil Gas Chromatography	d18O and dD Isotopes of Water	RSK 175 Dissolved Gas Quantification
2	FAC 5 GAS LINE PRODUCT	3.26.26	09:40	OIL									
2	FAC 5 GAS LINE PRODUCT	3.26.26	09:42	OIL + WATER*							<input checked="" type="checkbox"/>		

Chain of Custody Record Comments: **P10N VOAS (OIL) 2-402 pcs provided as extra if needed*

Relinquished by Signature	Company	Date	Time	Received by Signature	Company	Date	Time
<i>[Signature]</i>	ECMC	3.26.26	10:56	<i>[Signature]</i>	DIG	3/26/26	10:56

*Gas composition vs RSK-175 - Gas composition is a basic analysis of the concentration (ppm) of gases within the headspace of the sample (headspace is created at the lab). RSK-175 is a specific analysis technique combined with calculations to give the total dissolved gas of each species in the water sample (mg/L). Why one or the other? Gas composition gives us a quick, general look at relative concentrations and ratios (e.g., gas wetness). RSK-175 gives us an exact total of gas present in the sample (headspace and dissolved in the water). Questions? Give us a call at 303-531-2030.

** Rush and Expedited Rush turnaround time analysis will incur additional costs at 2x and 3x the standard turnaround time pricing.
* *WHOLE OIL GAS CHROMATOGRAPHY w/ PHYTAN & PRISTANE*