

Lab #: 430227 Job #: 25074 IS-70996 Co. Job#: _____
 Sample Name: Morton-1 Co. Lab#: _____
 Company: Colorado Oil & Gas Conservation
 API/Well: _____
 Container: IsoFlask
 Field/Site Name: COGCC: Morton Well
 Location: Brighton, CO
 Formation/Depth: _____
 Sampling Point: _____
 Date Sampled: 4/25/2014 Date Received: 4/30/2014 Date Reported: 6/11/2014

Component	Chemical mol. %	$\delta^{13}\text{C}$ ‰	δD ‰	$\delta^{18}\text{O}$ ‰
Carbon Monoxide -----	nd			
Helium -----	na			
Hydrogen -----	nd			
Argon -----	0.553			
Oxygen -----	0.090			
Nitrogen -----	21.88			
Carbon Dioxide -----	6.57			
Methane -----	59.51	-51.96	-237.3	
Ethane -----	7.88	-32.01		
Ethylene -----	nd			
Propane -----	2.49	-27.35		
Propylene -----	nd			
Iso-butane -----	0.295			
N-butane -----	0.501	-26.46		
Iso-pentane -----	0.120			
N-pentane -----	0.0737			
Hexanes + -----	0.0373			

Remarks:

Analysis is of gas extracted from water by headspace equilibration. Analysis has been corrected for helium added to create headspace. Helium dilution factor = 0.59
 Concentration of methane in water = 41 cc/L ; 28 ppm
 Concentration of ethane in water = 5.9 cc/L ; 7.3 ppm
 Concentration of propane in water = 1.8 cc/L ; 3.2 ppm
 *Addition of helium negates the ability to detect native helium and may negate the ability to detect hydrogen.

nd = not detected. na = not analyzed. Isotopic composition of hydrogen is relative to VSMOW. Isotopic composition of carbon is relative to VPDB. Isotopic composition of oxygen is relative to VSMOW, except for carbon dioxide which is relative to VPDB. Calculations for BTU and specific gravity per ASTM D3588. Chemical compositions are normalized to 100%. Mol. % is approximately equal to vol. %.